

# Package ‘photosynthesis’

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**Title** Model C3 Photosynthesis

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**Imports** checkmate (>= 1.9.0), crayon (>= 1.3.0), dplyr (>= 0.8.0), furrr (>= 0.1.0), future (>= 1.10.0), glue (>= 1.3.0), gunit (>= 1.0.0), magrittr (>= 1.5.0), methods (>= 3.5.0), purrr (>= 0.2.5), rlang (>= 0.3.0), stringr (>= 1.3.0), tealeaves (>= 1.0.5), tidyselect (>= 0.2.5)

**Suggests** ggplot2, knitr, rmarkdown, tidyr, testthat

**Description** Simulate C3 photosynthesis using the Farquhar, von Caemmerer, Berry (1980) <doi:10.1007/BF00386231> model as described in Buckley and Diaz-Espejo (2015) <doi:10.1111/pce.12459>. It uses units to ensure that parameters are properly specified and transformed before calculations. Temperature response functions get automatically ``baked'' into all parameters based on leaf temperature following Bernacchi et al. (2002) <doi:10.1104/pp.008250>. The package includes boundary layer, cuticular, stomatal, and mesophyll conductances to CO<sub>2</sub>, which each can vary on the upper and lower portions of the leaf. Use straightforward functions to simulate photosynthesis over environmental gradients such as Photosynthetic Photon Flux Density (PPFD) and leaf temperature, or over trait gradients such as CO<sub>2</sub> conductance or photochemistry.

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**Author** Chris Muir [aut, cre, cph] (<<https://orcid.org/0000-0003-2555-3878>>)

**Maintainer** Chris Muir <[cdmuir@hawaii.edu](mailto:cdmuir@hawaii.edu)>

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photosynthesis-package  
*photosynthesis package*

### Description

Model C3 Photosynthesis

### Details

See the README on [GitHub](#)

A\_supply

*CO2 supply and demand function (mol / m^2 s)*

### Description

This function is not intended to be called by users directly.

### Usage

```
A_supply(C_chl, pars, unitless = FALSE)

A_demand(C_chl, pars, unitless = FALSE)
```

**Arguments**

C_chl	Chloroplastic CO2 concentration in Pa of class units
pars	Concatenated parameters (leaf_par, enviro_par, and constants)
unitless	Logical. Should units be set? The function is faster when FALSE, but input must be in correct units or else results will be incorrect without any warning.

**Details****Supply function:**

$$A = g_{tc}(C_{air} - C_{chl})$$

**Demand function:**

$$A = (1 - \Gamma * /C_{chl})\min(W_{carbox}, W_{regen}, W_{tpu}) - R_d$$

Symbol	R	Description	Units	Default
A	A	photosynthetic rate	$\mu\text{mol CO}_2 / (\text{m}^2 \text{s})$	calculated
g_tc	g_tc	total conductance to CO2	$\mu\text{mol CO}_2 / (\text{m}^2 \text{s Pa})$	calculated
C_air	C_air	atmospheric CO2 concentration	Pa	41
C_chl	C_chl	chloroplastic CO2 concentration	Pa	calculated
R_d	R_d	nonphotorespiratory CO2 release	$\mu\text{mol CO}_2 / (\text{m}^2 \text{s})$	2
$\Gamma*$	gamma_star	chloroplastic CO2 compensation point	Pa	3.743

**Value**

Value in mol / ( $\text{m}^2 \text{s}$ ) of class units

**Examples**

```
bake_par <- make_bakepar()
constants <- make_constants(use_tealeaves = FALSE)
enviro_par <- make_enviropar(use_tealeaves = FALSE)
leaf_par <- make_leafpar(use_tealeaves = FALSE)
leaf_par <- bake(leaf_par, bake_par, constants)
# Or bake with piping (need library(magrittr))
# leaf_par %>% bake(bake_par, constants)
enviro_par$T_air <- leaf_par$T_leaf

pars <- c(leaf_par, enviro_par, constants)
C_chl <- set_units(35, "Pa")

A_supply(C_chl, pars)

A_demand(C_chl, pars)
```

bake

*Leaf parameter temperature responses*

## Description

'bake' leaf parameters using temperature response functions

Constructor function for baked class. This will also inherit class [leaf\\_par](#). This function ensures that temperature is "baked in" to leaf parameter calculations T\_leaf using temperature response functions detailed below.

## Usage

```
bake(leaf_par, bake_par, constants, assert_units = TRUE)

temp_resp1(par25, E_a, R, T_leaf, T_ref, unitless)

temp_resp2(par25, D_s, E_a, E_d, R, T_leaf, T_ref, unitless)
```

## Arguments

<code>leaf_par</code>	A list of leaf parameters inheriting class <a href="#">leaf_par</a> . This can be generated using the <a href="#">make_leafpar</a> function.
<code>bake_par</code>	A list of temperature response parameters inheriting class <a href="#">bake_par</a> . This can be generated using the <a href="#">make_bakepar</a> function.
<code>constants</code>	A list of physical constants inheriting class <a href="#">constants</a> . This can be generated using the <a href="#">make_constants</a> function.
<code>assert_units</code>	Logical. Should parameter units be checked? The function is faster when FALSE, but input must be in correct units or else results will be incorrect without any warning.
<code>par25</code>	Parameter value at 25 °C of class units.
<code>E_a</code>	Empirical temperature response value in J/mol of class units.
<code>R</code>	Ideal gas constant in J / (mol K) of class units. See <a href="#">make_constants</a> .
<code>T_leaf</code>	Leaf temperature in K of class units. Will be converted to °C.
<code>T_ref</code>	Reference temperature in K of class units.
<code>unitless</code>	Logical. Should units be set? The function is faster when FALSE, but input must be in correct units or else results will be incorrect without any warning.
<code>D_s</code>	Empirical temperature response value in J / (mol K) of class units.
<code>E_d</code>	Empirical temperature response value in J/mol of class units.

## Details

Several leaf parameters ([leaf\\_par](#)) are temperature sensitive. Temperature-sensitive parameters are input at a reference temperature of 25 °C. These parameters are provided as par\_name25 and then "baked" using the appropriate temperature response function and parameters in [bake\\_par](#). The "baked" parameter will have the name without "25" appended (par\_name). E.g. V\_cmax25 becomes V\_cmax.

Temperature response functions following Bernacchi et al. 2002.

Temperature response function 1 ([temp\\_response1](#)):

$$\text{par}(T_{\text{leaf}}) = \text{par25} \exp(E_a/(RT_{\text{ref}})(T_{\text{leaf}} - 25)/(T_{\text{leaf}} + 273.15))$$

$T_{\text{ref}}$  is the reference temperature in K

$T_{\text{leaf}}$  is the leaf temperature in °C

Temperature response function 2 ([temp\\_response2](#)) is the above equation multiplied by:

$$(1 + \exp((D_s/R - E_d/(RT_{\text{ref}}))))/(1 + \exp((D_s/R) - (E_d/(R(T_{\text{leaf}} + 273.15)))))$$

Function 1 increases exponentially with temperature; Function 2 peaks a particular temperature.

## References

Bernacchi CJ, Portis AR, Nakano H, von Caemmerer S, Long SP. 2002. Temperature response of mesophyll conductance. Implications for the determination of Rubisco enzyme kinetics and for limitations to photosynthesis in vivo. *Plant Physiology* 130: 1992-8.

## Examples

```
bake_par <- make_bakepar()
constants <- make_constants(use_tealeaves = FALSE)
leaf_par <- make_leafpar(replace = list(T_leaf = set_units(293.15, "K")),
                           use_tealeaves = FALSE)
baked_leafpar <- bake(leaf_par, bake_par, constants)

baked_leafpar$V_cmax25
baked_leafpar$V_cmax
```

## Description

See [bake](#)

**bake\_par***S3 class bake\_par***Description**

Constructor function for **bake\_par** class. This function ensures that leaf temperature gets properly "baked" into leaf parameters.

**Usage**

```
bake_par(.x)
```

**Arguments**

.x	A list to be constructed into <b>bake_par</b> .
----	---

**CO2\_conductance***Conductance to CO2 (umol / (m^2 s Pa))***Description**

Conductance to CO2 (umol / (m<sup>2</sup> s Pa))

- g\_tc: total conductance to CO2
- g\_uc: cuticular conductance to CO2
- g\_bc: boundary layer conductance to CO2
- g\_mc: mesophyll conductance to CO2
- g\_sc: stomatal conductance to CO2

**Usage**

```
.get_gtc(pars, unitless)
.get_guc(pars, surface, unitless)
.get_gbc(pars, surface, unitless)
.get_gmc(pars, surface, unitless)
.get_gsc(pars, surface, unitless)
```

## Arguments

pars	Concatenated parameters (leaf_par, enviro_par, and constants)
unitless	Logical. Should units be set? The function is faster when FALSE, but input must be in correct units or else results will be incorrect without any warning.
surface	Leaf surface (lower or upper)

## Details

Total conductance to CO<sub>2</sub> is the sum of parallel conductances on the lower ( $g_{c,\text{lower}}$ ) and upper ( $g_{c,\text{upper}}$ ) leaf portions:

$$g_{c,\text{total}} = g_{c,\text{lower}} + g_{c,\text{upper}}$$

Each partial conductance consists of two parallel conductances, the cuticular conductance ( $g_{u,c}$ ) and the in-series conductances through mesophyll ( $g_{m,c}$ ), stomata ( $g_{s,c}$ ), and boundary layer ( $g_{b,c}$ ). To simplify the formula, I use substitute resistance where  $r_x = 1/g_x$ . For surface  $i$ :

$$g_{c,i} = g_{u,i} + (1/(r_{m,i} + r_{s,i} + r_{b,i}))$$

The cuticular, stomatal, and mesophyll conductances can be the same or different for upper and lower. The partitioning factors ( $k_x$ ) divide the conductance between surfaces while keeping the total conductance constant:

$$g_{x,\text{lower}} = g_x(1/(1 + k_x))$$

$$g_{x,\text{upper}} = g_x(k_x/(1 + k_x))$$

$$g_x = g_{x,\text{lower}} + g_{x,\text{upper}}$$

How the partitioning factors work:

$k_x$	description
0	all conductance on <b>lower</b> surface/portion
0.5	2/3 conductance on <b>lower</b> surface
1	conductance evenly divided between surfaces/portions
2	2/3 conductance on <b>upper</b> surface
Inf	all conductance on <b>upper</b> surface/portion

The boundary layer conductances for each are calculated on the basis of mass and heat transfer (see [.get\\_gbc](#)).

Symbol	R	Description	Units	Default
$g_{mc}$	$g_{\text{mc}}$	mesophyll conductance to CO <sub>2</sub> (T_leaf)	$\mu\text{mol CO}_2 / (\text{m}^2 \text{ s Pa})$	calculated
$g_{sc}$	$g_{\text{sc}}$	stomatal conductance to CO <sub>2</sub>	$\mu\text{mol CO}_2 / (\text{m}^2 \text{ s Pa})$	4
$g_{uc}$	$g_{\text{uc}}$	cuticular conductance to CO <sub>2</sub>	$\mu\text{mol CO}_2 / (\text{m}^2 \text{ s Pa})$	0.1
$k_{mc}$	$k_{\text{mc}}$	partition of $g_{mc}$ to lower mesophyll	none	1

$k_{sc}$	<code>k_sc</code>	partition of $g_{sc}$ to lower surface	none	1
$k_{uc}$	<code>k_uc</code>	partition of $g_{uc}$ to lower surface	none	1

**constants***S3 class constants***Description**

Constructor function for constants class. This function ensures that physical constant inputs are properly formatted.

**Usage**

```
constants(.x, use_tealeaves)
```

**Arguments**

- .x                    A list to be constructed into **constants**.
- use\_tealeaves      Logical. Should leaf energy balance be used to calculate leaf temperature (T\_leaf)? If TRUE, `tleaf` calculates T\_leaf. If FALSE, user-defined T\_leaf is used. Additional parameters and constants are required, see [make\\_parameters](#).

**enviro\_par***S3 class enviro\_par***Description**

Constructor function for enviro\_par class. This function ensures that environmental parameter inputs are properly formatted.

**Usage**

```
enviro_par(.x, use_tealeaves)
```

**Arguments**

- .x                    A list to be constructed into **enviro\_par**.
- use\_tealeaves      Logical. Should leaf energy balance be used to calculate leaf temperature (T\_leaf)? If TRUE, `tleaf` calculates T\_leaf. If FALSE, user-defined T\_leaf is used. Additional parameters and constants are required, see [make\\_parameters](#).

---

FvCB*Farquhar-von Caemmerer-Berry (FvCB) C3 photosynthesis model*

---

**Description**

Farquhar-von Caemmerer-Berry (FvCB) C3 photosynthesis model  
 Rubisco-limited assimilation rate  
 RuBP regeneration-limited assimilation rate  
 TPU-limited assimilation rate

**Usage**

```
FvCB(C_chl, pars, unitless = FALSE)

W_carbox(C_chl, pars, unitless = FALSE)

W_regen(C_chl, pars, unitless = FALSE)

W_tpu(C_chl, pars, unitless = FALSE)
```

**Arguments**

C_chl	Chloroplastic CO <sub>2</sub> concentration in Pa of class units
pars	Concatenated parameters (leaf_par, enviro_par, and constants)
unitless	Logical. Should units be set? The function is faster when FALSE, but input must be in correct units or else results will be incorrect without any warning.

**Details**

Equations following Buckley and Diaz-Espejo (2015):

**Rubisco-limited assimilation rate:**

$$W_{\text{carbox}} = V_{c,\text{max}} C_{\text{chl}} / (C_{\text{chl}} + K_m)$$

where:

$$K_m = K_C (1 + O/K_O)$$

**RuBP regeneration-limited assimilation rate:**

$$W_{\text{regen}} = J C_{\text{chl}} / (4 C_{\text{chl}} + 8 \Gamma *)$$

where  $J$  is a function of PPF, obtained by solving the equation:

$$0 = \theta_J J^2 - J(J_{\max} + \phi_J PPF D) + J_{\max} \phi_J PPF D$$

### TPU-limited assimilation rate:

$$W_{\text{tpu}} = 3V_{\text{tpu}}C_{\text{chl}}/(C_{\text{chl}} - \Gamma*)$$

Symbol	R	Description	Units	Default
$C_{\text{chl}}$	$C_{\text{chl}}$	chloroplastic CO <sub>2</sub> concentration	Pa	input
$\Gamma*$	gamma_star	chloroplastic CO <sub>2</sub> compensation point (T_leaf)	Pa	calculated
$J_{\max}$	J_max	potential electron transport (T_leaf)	$\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$	calculated
$K_C$	K_C	Michaelis constant for carboxylation (T_leaf)	$\mu\text{mol / mol}$	calculated
$K_O$	K_O	Michaelis constant for oxygenation (T_leaf)	$\mu\text{mol / mol}$	calculated
$O$	O	atmospheric O <sub>2</sub> concentration	kPa	21.27565
$\phi_J$	phi_J	initial slope of the response of J to PPFD	none	0.331
PPFD	PPFD	photosynthetic photon flux density	$\mu\text{mol quanta / (m}^2 \text{ s)}$	1500
$R_d$	R_d	nonphotorespiratory CO <sub>2</sub> release (T_leaf)	$\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$	calculated
$\theta_J$	theta_J	curvature factor for light-response curve	none	0.825
$V_{c,\max}$	V_cmax	maximum rate of carboxylation (T_leaf)	$\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$	calculated
$V_{\text{tpu}}$	V_tpu	rate of triose phosphate utilization (T_leaf)	$\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$	calculated

### Value

A list of four values with units  $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$  of class `units`:

- $W_{\text{carbox}}$ : Rubisco-limited assimilation rate
- $W_{\text{regen}}$ : RuBP regeneration-limited assimilation rate
- $W_{\text{tpu}}$ : TPU-limited assimilation rate
- A: minimum of  $W_{\text{carbox}}$ ,  $W_{\text{regen}}$ , and  $W_{\text{tpu}}$

### References

- Buckley TN and Diaz-Espejo A. 2015. Partitioning changes in photosynthetic rate into contributions from different variables. *Plant, Cell & Environment* 38: 1200-11.
- Farquhar GD, Caemmerer S, Berry JA. 1980. A biochemical model of photosynthetic CO<sub>2</sub> assimilation in leaves of C3 species. *Planta* 149: 78–90.

### Examples

```
bake_par <- make_bakepar()
constants <- make_constants(use_tealeaves = FALSE)
enviro_par <- make_enviropar(use_tealeaves = FALSE)
leaf_par <- make_leafpar(use_tealeaves = FALSE)
leaf_par <- bake(leaf_par, bake_par, constants)
```

```

pars <- c(leaf_par, enviro_par, constants)
C_chl <- set_units(24.28, "Pa")
FvCB(C_chl, pars)

```

gw2gc

*Convert g\_c ( $\mu\text{mol CO}_2/\text{m}^2/\text{s}/\text{Pa}$ ) to g\_w ( $\mu\text{mol H}_2\text{O}/\text{m}^2/\text{s}/\text{Pa}$ )*

## Description

Convert g\_c ( $\mu\text{mol CO}_2/\text{m}^2/\text{s}/\text{Pa}$ ) to g\_w ( $\mu\text{mol H}_2\text{O}/\text{m}^2/\text{s}/\text{Pa}$ )  
 Convert g\_c (umol CO2/m^2/s/Pa) to g\_w (umol H2O /m^2/s/Pa)

## Usage

```
gw2gc(g_w, D_c, D_w, unitless)

gc2gw(g_c, D_c, D_w, unitless)
```

## Arguments

g_w	conductance to water vapor in units ( $\mu\text{mol H}_2\text{O}/(\text{m}^2 \text{s Pa})$ ) of class units.
D_c	diffusion coefficient for CO2 in air in units of $\text{m}^2/\text{s}$ of call units
D_w	diffusion coefficient for H2O in air in units of $\text{m}^2/\text{s}$ of call units
unitless	Logical. Should units be set? The function is faster when FALSE, but input must be in correct units or else results will be incorrect without any warning.
g_c	conductance to CO2 in units ( $\mu\text{mol H}_2\text{O}/(\text{m}^2 \text{s Pa})$ ) of class units.

## Details

Diffusive conductance to CO2 is generally about ~1.6x that of H2O because of the higher molecular weight. To convert, multiply conductance by the ratio of diffusion coefficients:

$$g_c = g_w D_c / D_w$$

$$g_w = g_c D_w / D_c$$

## Value

Value with units  $\mu\text{mol}/(\text{m}^2 \text{s Pa})$  of class units.

## Note

This function will soon be moving to the standalone gunit package.

## Examples

```
D_c <- set_units(1.29e-05, "m^2/s")
D_w <- set_units(2.12e-05, "m^2/s")
g_c <- set_units(3, "umol/m^2/s/Pa")
g_w <- gc2gw(g_c, D_c, D_w, unitless = FALSE)
g_w

gw2gc(g_w, D_c, D_w, unitless = FALSE)
```

J

*J: Rate of electron transport (umol/m^2/s)*

## Description

Calculate the rate of electron transport as a function of photosynthetic photon flux density (PPFD).

## Usage

```
J(pars, unitless = FALSE)
```

## Arguments

- |          |  |
|----------|--|
| pars     | Concatenated parameters (leaf_par, enviro_par, and constants)  |
| unitless | Logical. Should units be set? The function is faster when FALSE, but input must be in correct units or else results will be incorrect without any warning. |

## Details

*J* as a function of PPFD is the solution to the quadratic expression:

$$0 = \theta_J J^2 - J(J_{\max} + \phi_J \text{PPFD}) + J_{\max}\phi_J \text{PPFD}$$

Symbol	R	Description	Units	Default
$J_{\max}$	J_max	potential electron transport (T_leaf)	$\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$	calculated
$\phi_J$	phi_J	initial slope of the response of J to PPFD	none	0.331
PPFD	PPFD	photosynthetic photon flux density	$\mu\text{mol quanta} / (\text{m}^2 \text{ s})$	1500
$\theta_J$	theta_J	curvature factor for light-response curve	none	0.825

## Value

Value in  $\mu\text{mol}/(\text{m}^2 \text{ s})$  of class units

## Examples

```

library(magrittr)
library(photosynthesis)

bake_par <- make_bakepar()
constants <- make_constants(use_tealeaves = FALSE)
enviro_par <- make_enviropar(use_tealeaves = FALSE)
leaf_par <- make_leafpar(use_tealeaves = FALSE)
enviro_par$T_air <- leaf_par$T_leaf
leaf_par %>>% bake(bake_par, constants)

pars <- c(leaf_par, enviro_par, constants)
J(pars, FALSE)

```

**leaf\_par***S3 class leaf\_par***Description**

Constructor function for leaf\_par class. This function ensures that leaf parameter inputs are properly formatted.

**Usage**

```
leaf_par(.x, use_tealeaves)
```

**Arguments**

- |               |  |
|---------------|--|
| .x            | A list to be constructed into <b>leaf_par</b> .  |
| use_tealeaves | Logical. Should leaf energy balance be used to calculate leaf temperature (T_leaf)? If TRUE, <a href="#">tleaf</a> calculates T_leaf. If FALSE, user-defined T_leaf is used. Additional parameters and constants are required, see <a href="#">make_parameters</a> . |

**make\_parameters***Make lists of parameters for photosynthesis***Description**

Make lists of parameters for photosynthesis

[make\\_leafpar](#)  
[make\\_enviropar](#)  
[make\\_bakepar](#)  
[make\\_constants](#)

**Usage**

```
make_leafpar(replace = NULL, use_tealeaves)

make_enviropar(replace = NULL, use_tealeaves)

make_bakepar(replace = NULL)

make_constants(replace = NULL, use_tealeaves)
```

**Arguments**

<code>replace</code>	A named list of parameters to replace defaults. If <code>NULL</code> , defaults will be used.
<code>use_tealeaves</code>	Logical. Should leaf energy balance be used to calculate leaf temperature ( <code>T_leaf</code> )? If <code>TRUE</code> , <code>tleaf</code> calculates <code>T_leaf</code> . If <code>FALSE</code> , user-defined <code>T_leaf</code> is used. Additional parameters and constants are required, see <a href="#">make_parameters</a> .

**Details****Constants:**

<i>Symbol</i>	<i>R</i>	<i>Description</i>	<i>Units</i>	<i>Default</i>
$D_{c,0}$	<code>D_c0</code>	diffusion coefficient for CO <sub>2</sub> in air at 0 °C	m <sup>2</sup> / s	1.29e-5
$D_{h,0}$	<code>D_h0</code>	diffusion coefficient for heat in air at 0 °C	m <sup>2</sup> / s	1.90e-5
$D_{m,0}$	<code>D_m0</code>	diffusion coefficient for momentum in air at 0 °C	m <sup>2</sup> / s	1.33e-5
$D_{w,0}$	<code>D_w0</code>	diffusion coefficient for water vapor in air at 0 °C	m <sup>2</sup> / s	2.12e-5
$\epsilon$	<code>epsilon</code>	ratio of water to air molar masses	none	0.622
$G$	<code>G</code>	gravitational acceleration	m / s <sup>2</sup>	9.8
$eT$	<code>eT</code>	exponent for temperature dependence of diffusion	none	1.75
$R$	<code>R</code>	ideal gas constant	J / (mol K)	8.3144598
$\sigma$	<code>s</code>	Stephan-Boltzmann constant	W / (m <sup>2</sup> K <sup>4</sup> )	5.67e-08
$Sh$	<code>Sh</code>	Sherwood number	none	<a href="#">calculated</a>

**Baking (i.e. temperature response) parameters:**

<i>Symbol</i>	<i>R</i>	<i>Description</i>	<i>Units</i>	<i>Default</i>
$D_{s,gmc}$	<code>Ds_gmc</code>	empirical temperature response parameter	J / (mol K)	487.29
$D_{s,Jmax}$	<code>Ds_Jmax</code>	empirical temperature response parameter	J / (mol K)	388.04
$E_{a,\Gamma^*}$	<code>Ea_gammastar</code>	empirical temperature response parameter	J / mol	24459.97
$E_{a,gmc}$	<code>Ea_gmc</code>	empirical temperature response parameter	J / mol	68901.56
$E_{a,Jmax}$	<code>Ea_Jmax</code>	empirical temperature response parameter	J / mol	56095.18
$E_{a,KC}$	<code>Ea_KC</code>	empirical temperature response parameter	J / mol	80989.78
$E_{a,KO}$	<code>Ea_KO</code>	empirical temperature response parameter	J / mol	23719.97
$E_{a,Rd}$	<code>Ea_Rd</code>	empirical temperature response parameter	J / mol	40446.75
$E_{a,Vcmax}$	<code>Ea_Vcmax</code>	empirical temperature response parameter	J / mol	52245.78
$E_{d,gmc}$	<code>Ed_gmc</code>	empirical temperature response parameter	J / mol	148788.56
$E_{d,Jmax}$	<code>Ed_Jmax</code>	empirical temperature response parameter	J / mol	121244.79

**Environment parameters:**

<i>Symbol</i>	<i>R</i>	<i>Description</i>	<i>Units</i>	<i>Default</i>
$C_{\text{air}}$	$C_{\text{air}}$	atmospheric CO <sub>2</sub> concentration	Pa	41
$O$	$O$	atmospheric O <sub>2</sub> concentration	kPa	21.27565
$P$	$P$	atmospheric pressure	kPa	101.3246
PPFD	PPFD	photosynthetic photon flux density	umol quanta / (m <sup>2</sup> s)	1500
RH	RH	relative humidity	none	0.50
$u$	wind	windspeed	m / s	2

**Leaf parameters:**

<i>Symbol</i>	<i>R</i>	<i>Description</i>	<i>Units</i>	<i>Default</i>
$d$	leafsize	leaf characteristic dimension	m	0.1
$\Gamma^*$	gamma_star	chloroplastic CO <sub>2</sub> compensation point (T_leaf)	Pa	calculated
$\Gamma^{*25}$	gamma_star25	chloroplastic CO <sub>2</sub> compensation point (25 °C)	Pa	3.743
$g_{\text{mc}}$	$g_{\text{mc}}$	mesophyll conductance to CO <sub>2</sub> (T_leaf)	μmol CO <sub>2</sub> / (m <sup>2</sup> s Pa)	calculated
$g_{\text{mc}}$	$g_{\text{mc}25}$	mesophyll conductance to CO <sub>2</sub> (25 °C)	μmol CO <sub>2</sub> / (m <sup>2</sup> s Pa)	4
$g_{\text{sc}}$	$g_{\text{sc}}$	stomatal conductance to CO <sub>2</sub>	μmol CO <sub>2</sub> / (m <sup>2</sup> s Pa)	4
$g_{\text{uc}}$	$g_{\text{uc}}$	cuticular conductance to CO <sub>2</sub>	μmol CO <sub>2</sub> / (m <sup>2</sup> s Pa)	0.1
$J_{\text{max}25}$	$J_{\text{max}25}$	potential electron transport (25 °C)	μmol CO <sub>2</sub> / (m <sup>2</sup> s)	200
$J_{\text{max}}$	$J_{\text{max}}$	potential electron transport (T_leaf)	μmol CO <sub>2</sub> / (m <sup>2</sup> s)	calculated
$k_{\text{mc}}$	$k_{\text{mc}}$	partition of $g_{\text{mc}}$ to lower mesophyll	none	1
$k_{\text{sc}}$	$k_{\text{sc}}$	partition of $g_{\text{sc}}$ to lower surface	none	1
$k_{\text{uc}}$	$k_{\text{uc}}$	partition of $g_{\text{uc}}$ to lower surface	none	1
$K_{\text{C}25}$	$K_{\text{C}25}$	Michaelis constant for carboxylation (25 °C)	μmol / mol	268.3
$K_{\text{C}}$	$K_{\text{C}}$	Michaelis constant for carboxylation (T_leaf)	μmol / mol	calculated
$K_{\text{O}25}$	$K_{\text{O}25}$	Michaelis constant for oxygenation (25 °C)	μmol / mol	165084.2
$K_{\text{O}}$	$K_{\text{O}}$	Michaelis constant for oxygenation (T_leaf)	μmol / mol	calculated
$\phi_J$	phi_J	initial slope of the response of J to PPFD	none	0.331
$R_{\text{d}25}$	$R_{\text{d}25}$	nonphotorespiratory CO <sub>2</sub> release (25 °C)	μmol CO <sub>2</sub> / (m <sup>2</sup> s)	2
$R_{\text{d}}$	$R_{\text{d}}$	nonphotorespiratory CO <sub>2</sub> release (T_leaf)	μmol CO <sub>2</sub> / (m <sup>2</sup> s)	calculated
$\theta_J$	theta_J	curvature factor for light-response curve	none	0.825
$T_{\text{leaf}}$	$T_{\text{leaf}}$	leaf temperature	K	298.15
$V_{\text{c},\text{max}25}$	$V_{\text{cmax}25}$	maximum rate of carboxylation (25 °C)	μmol CO <sub>2</sub> / (m <sup>2</sup> s)	150
$V_{\text{c},\text{max}}$	$V_{\text{cmax}}$	maximum rate of carboxylation (T_leaf)	μmol CO <sub>2</sub> / (m <sup>2</sup> s)	calculated
$V_{\text{tpu}25}$	$V_{\text{tpu}25}$	rate of triose phosphate utilization (25 °C)	μmol CO <sub>2</sub> / (m <sup>2</sup> s)	200
$V_{\text{tpu}}$	$V_{\text{tpu}}$	rate of triose phosphate utilisation (T_leaf)	μmol CO <sub>2</sub> / (m <sup>2</sup> s)	calculated

If use\_tealeaves = TRUE, additional parameters are:

**Constants:**

<i>Symbol</i>	<i>R</i>	<i>Description</i>	<i>Units</i>	<i>Default</i>
$c_p$	$c_p$	heat capacity of air	J / (g K)	1.01
$R_{\text{air}}$	$R_{\text{air}}$	specific gas constant for dry air	J / (kg K)	287.058

**Environmental parameters:**

<i>Symbol</i>	<i>R</i>	<i>Description</i>	<i>Units</i>	<i>Default</i>
---------------	----------	--------------------	--------------	----------------

$E_q$	$E_{\text{q}}$	energy per mole quanta	$\text{kJ} / \text{mol}^2$	22
$f_{\text{PAR}}$	$f_{\text{par}}$	fraction of incoming shortwave radiation that is photosynthetically active radiation (PAR)	none	0.5
$r$	$r$	reflectance for shortwave irradiance (albedo)	none	0.2
$T_{\text{air}}$	$T_{\text{air}}$	air temperature	K	29

**Leaf parameters:**

<i>Symbol</i>	<i>R</i>	<i>Description</i>	<i>Units</i>	<i>Default</i>
$\alpha_l$	$\text{abs\_l}$	absorbivity of longwave radiation (4 - 80 $\mu\text{m}$ )	none	0.97
$\alpha_s$	$\text{abs\_s}$	absorbivity of shortwave radiation (0.3 - 4 $\mu\text{m}$ )	none	0.50
$g_{\text{sw}}$	$g_{\text{sw}}$	stomatal conductance to H <sub>2</sub> O	( $\mu\text{mol H}_2\text{O}$ ) / ( $\text{m}^2 \text{s Pa}$ )	converted from $g_{\text{sc}}$
$g_{\text{uw}}$	$g_{\text{uw}}$	cuticular conductance to H <sub>2</sub> O	( $\mu\text{mol H}_2\text{O}$ ) / ( $\text{m}^2 \text{s Pa}$ )	converted from $g_{\text{uc}}$
$\text{logit}(sr)$	$\text{logit\_sr}$	stomatal ratio (logit transformed)	none	converted from $k_{\text{sc}}$

**Value**

`make_leafpar`: An object inheriting from class `leaf_par`  
`make_enviropar`: An object inheriting from class `enviro_par`  
`make_bakepar`: An object inheriting from class `bake_par`  
`make_constants`: An object inheriting from class `constants`

**References**

Buckley TN and Diaz-Espejo A. 2015. Partitioning changes in photosynthetic rate into contributions from different variables. Plant, Cell & Environment 38: 1200-11.

**Examples**

```
bake_par <- make_bakepar()
constants <- make_constants(use_tealeaves = FALSE)
enviro_par <- make_enviropar(use_tealeaves = FALSE)
leaf_par <- make_leafpar(use_tealeaves = FALSE)

leaf_par <- make_leafpar(
  replace = list(
    g_sc = set_units(3, "umol/m^2/s/Pa"),
    V_cmax25 = set_units(100, "umol/m^2/s")
  ), use_tealeaves = FALSE
)
```

*parameter\_names*      *Get vector of parameter names*

**Description**

Get vector of parameter names

**Usage**

```
parameter_names(which, use_tealeaves)
```

**Arguments**

which	A character string indicating which parameter names to retrieve: "leaf", "enviro", "bake", or "constants". Partial matching allowed.
use_tealeaves	Logical. Should leaf energy balance be used to calculate leaf temperature ( $T_{leaf}$ )? If TRUE, <code>tleaf</code> calculates $T_{leaf}$ . If FALSE, user-defined $T_{leaf}$ is used. Additional parameters and constants are required, see <a href="#">make_parameters</a> .

**Examples**

```
parameter_names("leaf", use_tealeaves = FALSE)
```

---

photosynthesis	<i>Simulate C3 photosynthesis</i>
----------------	-----------------------------------

---

**Description**

`photosynthesis`: simulate C3 photosynthesis over multiple parameter sets  
`photo`: simulate C3 photosynthesis over a single parameter set

**Usage**

```
photosynthesis(  
  leaf_par,  
  enviro_par,  
  bake_par,  
  constants,  
  use_tealeaves,  
  progress = TRUE,  
  quiet = FALSE,  
  assert_units = TRUE,  
  parallel = FALSE  
)  
  
photo(  
  leaf_par,  
  enviro_par,  
  bake_par,  
  constants,  
  use_tealeaves,  
  quiet = FALSE,  
  assert_units = TRUE,
```

```

check = TRUE,
prepare_for_tleaf = use_tealeaves
)

```

## Arguments

<code>leaf_par</code>	A list of leaf parameters inheriting class <code>leaf_par</code> . This can be generated using the <code>make_leafpar</code> function.
<code>enviro_par</code>	A list of environmental parameters inheriting class <code>enviro_par</code> . This can be generated using the <code>make_enviropar</code> function.
<code>bake_par</code>	A list of temperature response parameters inheriting class <code>bake_par</code> . This can be generated using the <code>make_bakepar</code> function.
<code>constants</code>	A list of physical constants inheriting class <code>constants</code> . This can be generated using the <code>make_constants</code> function.
<code>use_tealeaves</code>	Logical. Should leaf energy balance be used to calculate leaf temperature ( <code>T_leaf</code> )? If TRUE, <code>tleaf</code> calculates <code>T_leaf</code> . If FALSE, user-defined <code>T_leaf</code> is used. Additional parameters and constants are required, see <a href="#">make_parameters</a> .
<code>progress</code>	Logical. Should a progress bar be displayed?
<code>quiet</code>	Logical. Should messages be displayed?
<code>assert_units</code>	Logical. Should parameter units be checked? The function is faster when FALSE, but input must be in correct units or else results will be incorrect without any warning.
<code>parallel</code>	Logical. Should parallel processing be used via <code>future_map</code> ?
<code>check</code>	Logical. Should arguments checks be done? This is intended to be disabled when <code>photo</code> is called from <code>photosynthesis</code> . Default is TRUE.
<code>prepare_for_tleaf</code>	Logical. Should arguments additional calculations for <code>tleaf</code> ? This is intended to be disabled when <code>photo</code> is called from <code>photosynthesis</code> . Default is <code>use_tealeaves</code> .

## Details

`photo`: This function takes simulates photosynthetic rate using the Farquhar-von Caemmerer-Berry (FvCB) model of C3 photosynthesis for single combined set of leaf parameters (`leaf_par`), environmental parameters (`enviro_par`), and physical constants (`constants`). Leaf parameters are provided at reference temperature (25 °C) and then "baked" to the appropriate leaf temperature using temperature response functions (see `bake`).

`photosynthesis`: This function uses `photo` to simulate photosynthesis over multiple parameter sets that are generated using `cross_df`.

## Value

A data.frame with the following units columns

### Input:

C_air	atmospheric CO2 concentration (Pa)
g_mc25	mesophyll conductance to CO2 at 25 °C ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s Pa})$ )
g_sc	stomatal conductance to CO2 ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s Pa})$ )
g_uc	cuticular conductance to CO2 ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s Pa})$ )
gamma_star25	chloroplastic CO2 compensation point at 25 °C (Pa)
J_max25	potential electron transport at 25 °C ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$ )
K_C25	Michaelis constant for carboxylation at 25 °C ( $\mu\text{mol / mol}$ )
K_O25	Michaelis constant for oxygenation at 25 °C ( $\mu\text{mol / mol}$ )
k_mc	partition of $g_{mc}$ to lower mesophyll (unitless)
k_sc	partition of $g_{sc}$ to lower surface (unitless)
k_uc	partition of $g_{uc}$ to lower surface (unitless)
leafsize	leaf characteristic dimension (m)
O	atmospheric O2 concentration (kPa)
P	atmospheric pressure (kPa)
phi_J	initial slope of the response of J to PPFD (unitless)
PPFD	photosynthetic photon flux density ( $\mu\text{mol quanta / (m}^2 \text{ s)}$ )
R_d25	nonphotorespiratory CO2 release at 25 °C ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$ )
RH	relative humidity (unitless)
theta_J	curvature factor for light-response curve (unitless)
T_air	air temperature (K)
T_leaf	leaf temperature (K)
V_cmax25	maximum rate of carboxylation at 25 °C ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$ )
V_tpu25	rate of triose phosphate utilization at 25 °C ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$ )
wind	wind speed (m / s)

**Baked Input:**

g_mc	mesophyll conductance to CO2 at T_leaf ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s Pa})$ )
gamma_star	chloroplastic CO2 compensation point at T_leaf (Pa)
J_max	potential electron transport at T_leaf ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$ )
K_C	Michaelis constant for carboxylation at T_leaf ( $\mu\text{mol / mol}$ )
K_O	Michaelis constant for oxygenation at T_leaf ( $\mu\text{mol / mol}$ )
R_d	nonphotorespiratory CO2 release at T_leaf ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$ )
V_cmax	maximum rate of carboxylation at T_leaf ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$ )
V_tpu	rate of triose phosphate utilisation at T_leaf ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$ )

**Output:**

A	photosynthetic rate at C_chl ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$ )
C_chl	chloroplastic CO2 concentration where A_supply intersects A_demand (Pa)
g_tc	total conductance to CO2 at T_leaf ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s Pa})$ )
value	A_supply - A_demand ( $\mu\text{mol CO}_2 / (\text{m}^2 \text{ s})$ ) at C_chl
convergence	convergence code (0 = converged)

**Examples**

```
# Single parameter set with 'photo'
```

```

bake_par <- make_bakepar()
constants <- make_constants(use_tealeaves = FALSE)
enviro_par <- make_enviropar(use_tealeaves = FALSE)
leaf_par <- make_leafpar(use_tealeaves = FALSE)
photo(leaf_par, enviro_par, bake_par, constants,
      use_tealeaves = FALSE)

# Multiple parameter sets with 'photosynthesis'

leaf_par <- make_leafpar(
  replace = list(
    T_leaf = set_units(c(293.14, 298.15), "K")
  ), use_tealeaves = FALSE
)
photosynthesis(leaf_par, enviro_par, bake_par, constants,
               use_tealeaves = FALSE)

```

ppm2pa

*Convert pressure from PPM to Pascals***Description**

Convert pressure from PPM to Pascals

**Usage**

```
ppm2pa(ppm, P)
```

**Arguments**

ppm	Pressure value in umol/mol of class units
P	Atmospheric pressure value in kPa of class units

**Details**

$$\text{Press}(kPa) = \text{Press}(ppm)P(kPa)$$

$$\text{Press}(Pa) = 1000\text{Press}(kPa)$$

**Value**

Value in Pa of class units

**Examples**

```

ppm <- set_units(400, "umol/mol")
P <- set_units(101.325, "kPa")
ppm2pa(ppm, P)

```

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